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G. Usha,^a S. Selvanayagam,^a M. Yogavel,^a D. Velmurugan,^a* A. Amalraj,^b R. Raghunathan,^b S. Shanmuga Sundara Raj^c and H.-K. Fun^d

^aDepartment of Crystallography and Biophysics, University of Madras, Guindy Campus, Chennai 600 025, India, ^bDepartment of Organic Chemistry, University of Madras, Guindy Campus, Chennai 600 025, India, ^cB3121 Medical Centre North, VanderBelt University, Department of Medicine-Nephrology, Nashville, USA, and ^dX-ray Crystallography Unit, School of Physics, Universiti Sains Malaysia, 11800 USM, Penang, Malaysia

Correspondence e-mail: d_velu@yahoo.com

Key indicators

Single-crystal X-ray study T = 293 KMean $\sigma(C-C) = 0.005 \text{ Å}$ R factor = 0.047 wR factor = 0.137 Data-to-parameter ratio = 10.8

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

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2-Benzoyl-3-cyclohexyl-4-(*p*-methoxyphenyl)-5-phenylpyrrolidine-3-spiro-3'chroman-4'-one

The title compound, $C_{38}H_{37}NO_4$, crystallizes with two molecules in the asymmetric unit. The molecules have similar geometric parameters and represent opposite enantiomers of the title compound. In both molecules, the pyrrolidine ring adopts an envelope conformation; the pyran ring in the chromanone moiety has a half-chair conformation. The molecular packing in the crystal is stabilized by $C-H\cdots O$ and $C-H\cdots \pi$ interactions, in addition to van der Waals forces.

Comment

Substituted pyrrolidines have gained much importance in recent years, since they are the basic structural elements of many alkaloids and pharmacologically active compounds (Jeyabharathi *et al.*, 2002). Pyrrolidine derivatives possess anti-influenza virus (Stylianakis *et al.*, 2003) and anticonvulsant (Obniska *et al.*, 2002) activity. The title compound, (I), a pyrrolidine derivative, was chosen for crystallographic study to determine its structure and function.



The asymmetric unit of (I) contains two molecules (Fig. 1); their corresponding bond lengths and bond angles are in good agreement.

All the C-C bond lengths in the aromatic rings (C, D, F and G) and the cyclohexyl ring (B) are comparable with the corresponding literature values of 1.384 (13) and 1.539 (16) Å, respectively (Allen *et al.*, 1987). The bond lengths in the pyrrolidine ring (A) are comparable with those reported earlier (Jeyabharathi *et al.*, 2001, 2002; Abdul Ajees *et al.*, 2002; Gzella & Wrzeciono, 1990). The bond lengths in the pyran ring (E) in the chromanone moiety are also close to those found in related structures (Jeyabharathi *et al.*, 2002; Abdul Ajees *et al.*, 2001).

The title compound has four chiral centres and crystallizes in the non-centrosymmetric space group $P2_12_12_1$. The lack of Received 18 August 2003 Accepted 22 August 2003 Online 30 September 2003



Figure 1

The molecular structure of the title compound, showing 30% probability displacement ellipsoids; H atoms have been omitted for clarity.



Figure 2 The C-H···O intramolecular interactions in molecule *A*...

effective anomalous scatterers did not allow us to determine the Flack (1983) parameter. It is noteworthy that the two independent molecules in the crystal structure of (I) are opposite enantiomers (Fig. 1, Table 1). The title compound is thus racemic, but crystallizes in a space group usually adopted by single enantiomers of chiral molecules.

The conformations of both molecules are quite similar. The dihedral angle between rings *C* and *D* is 56.5 (1)° for molecule *A* and 65.7 (1)° for molecule *B*. The dihedral angle between rings *F* and *G* is 30.7 (1)° for molecule *A* and 42.0 (1)° for molecule *B*.

The pyrrolidine ring (A) is in an envelope conformation with puckering parameters (Cremer & Pople, 1975) $q_2 =$ 0.370 (2) Å and $\varphi = -177.7$ (3)° for molecule A, and $q_2 =$ 0.409 (2) Å and $\varphi = 2.5$ (4)° for molecule B. Atom N1 deviates by 0.544 (2) and 0.598 (2) Å from the least-squares plane through the remaining four atoms (C2–C5) of that of ring for molecules A and B, respectively. The cyclohexyl ring adopts a chair conformation. This is confirmed by the puckering para-





meters $q_2 = 0.038$ (4) Å, $q_3 = 0.574$ (4) Å $= Q_T$, $\theta = 3.8$ (4)° and $\varphi = 10$ (7)° for molecule A, and $q_2 = 0.031$ (2) Å, $q_3 = -0.580$ (3) Å $= Q_T$, $\theta = 176.9$ (4)° and $\varphi = -165$ (5)° for molecule B. The pyran ring in the chromanone moiety has a half-chair conformation with the lowest asymmetry parameters of $\Delta C_2(C21-C20) = 0.059$ (2) for molecule A and 0.050 (1) for molecule B (Nardelli, 1983).

Intramolecular C-H···O interactions stabilize the conformation of the molecules (Fig. 2), whereas the crystal structure is stabilized by intermolecular C-H···O interactions (Table 2 and Fig. 3). In addition to this, the molecules in the crystal are also interlinked by C-H··· π interactions, so that H11*B* (bonded to C11*A*) is 2.83 Å from the centroid of phenyl ring *C* of molecule *B* transformed by $(\frac{3}{2} - x, 2 - y, z - \frac{1}{2})$, with a C11*A*-H11*B*···centroid angle of 158° and a C11*A*···centroid distance of 3.751 (4) Å; H11*D* (bonded to C11*B*) is 2.87 Å from the centroid of phenyl ring *C* of molecule *A* transformed by $(\frac{5}{2} - x, 2 - y, \frac{1}{2} + z)$, with a C11*B*-H11*D*···centroid angle of 152° and a C11*B*···centroid distance of 3.750 (3) Å.

Experimental

A mixture of (E)-3-(p-methylbenzylidene)-4-chromanone and *cis*-1-cyclohexyl-2-phenyl-3-benzoylaziridine was refluxed in toluene under a nitrogen atmosphere for 48 h. After the reaction was complete, the solvent was evaporated *in vacuo* and the resulting crude product was purified by column chromatography using a hexane–benzene mixture (7:3) as eluant. The product was crystallized from ethanol (Amalraj *et al.*, 1999).

Crystal data

$\begin{array}{l} C_{38} {\rm H}_{37} {\rm NO}_4 \\ M_r = 571.69 \\ Orthorhombic, \ P2_1 2_1 2_1 \\ a = 11.8945 \ (3) \ {\rm \AA} \\ b = 19.2698 \ (4) \ {\rm \AA} \\ c = 27.0279 \ (6) \ {\rm \AA} \\ V = 6194.9 \ (2) \ {\rm \AA}^3 \end{array}$	Mo $K\alpha$ radiation Cell parameters from 24284 reflections $\theta = 2.0-28.2^{\circ}$ $\mu = 0.08 \text{ mm}^{-1}$ T = 293 (2) K Block, yellow
Z = 8 $D_x = 1.226 \text{ Mg m}^{-3}$ Data collection	$0.24 \times 0.20 \times 0.16 \text{ mm}$
Siemens SMART CCD area- detector diffractometer ω scans Absorption correction: none 42360 measured reflections 8348 independent reflections	6500 reflections with $I > 2\sigma(I)$ $R_{int} = 0.037$ $\theta_{max} = 28.3^{\circ}$ $h = -10 \rightarrow 15$ $k = -25 \rightarrow 23$ $l = -35 \rightarrow 32$
Refinement	
Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.047$ $wR(F^2) = 0.137$	$w = 1/[\sigma^{2}(F_{o}^{2}) + (0.0668P)^{2} + 1.0304P]$ where $P = (F_{o}^{2} + 2F_{c}^{2})/3$

S = 1.078348 reflections 775 parameters H-atom parameters constrained

Table 1

Selected geometric parameters (Å, °).

1.461 (3)	N1B-C2B	1.464 (3)
1.464 (3)	N1B-C5B	1.465 (3)
1.557 (4)	C2B-C3B	1.559 (4)
1.562 (4)	C3B-C4B	1.564 (4)
1.576 (4)	C4B-C5B	1.568 (4)
1.536 (3)	C4B-C22B	1.530 (3)
1.219 (3)	C19B-O3B	1.212 (3)
1.477 (4)	C19B-C20B	1.484 (4)
1.395 (4)	C20B-C21B	1.385 (4)
1.409 (4)	C20B-C23B	1.403 (4)
1.365 (3)	C21B-O2B	1.374 (4)
1.435 (3)	C22B - O2B	1.438 (4)
-1140(2)	$C^{27}B - C^{3}B - C^{4}B - C^{19}B$	1154(2
116.9(2)	C2B-C3B-C4B-C19B	-115.3 (2
-96.3(2)	C19B-C4B-C5B-N1B	93.5 (2
141.4 (2)	C19B-C4B-C5B-C12B	-145.7 (2
-60.5(3)	C19B-C4B-C22B-O2B	62.4 (3
	$\begin{array}{c} 1.461 \ (3) \\ 1.464 \ (3) \\ 1.557 \ (4) \\ 1.557 \ (4) \\ 1.576 \ (4) \\ 1.576 \ (4) \\ 1.536 \ (3) \\ 1.219 \ (3) \\ 1.219 \ (3) \\ 1.477 \ (4) \\ 1.395 \ (4) \\ 1.395 \ (4) \\ 1.409 \ (4) \\ 1.365 \ (3) \\ 1.435 \ (3) \\ -114.0 \ (2) \\ -96.3 \ (2) \\ 141.4 \ (2) \\ -60.5 \ (3) \end{array}$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$

 $(\Delta/\sigma)_{\rm max} < 0.001$ $\Delta \rho_{\rm max} = 0.28$ e Å

 $\Delta \rho_{\rm min} = -0.17 \text{ e } \text{\AA}^{-3}$

Table 2

Hydrogen-bonding	geometry ((A, °	')	١.
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$D - H \cdots A$	$D-{\rm H}$	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
$C3A - H3A \cdots O3A$	0.98	2.26	2.813 (3)	115
$C3B - H3B \cdot \cdot \cdot O3B$	0.98	2.27	2.822 (3)	114
$C2A - H2A \cdots O1A$	0.98	2.49	3.028 (3)	114
$C2B - H2B \cdots O1B$	0.98	2.39	2.969 (3)	118
$C22A - H22B \cdots O1A$	0.97	2.53	3.120 (4)	119
$C22B - H22C \cdots O1B$	0.97	2.52	3.089 (4)	118
$C26A - H26A \cdots O1B^{i}$	0.93	2.59	3.310 (4)	134

Symmetry code: (i) x - 1, y, z.

The H atoms were positioned geometrically and were treated as riding on their parent C atoms, with aromatic C-H distances of 0.93 Å, methyl C-H distances of 0.96 Å, methine C-H distances of 0.98 Å and methylene C-H distances of 0.97 Å, and with $U_{iso} =$ $1.5U_{eq}(C)$ for methyl H atoms and $1.2U_{eq}(C)$ for other H atoms. Due to the lack of anomalous scatterers, the absolute structure was not determined from the X-ray diffraction data and Friedel pairs were merged.

Data collection: SMART (Siemens, 1996); cell refinement: SAINT (Siemens, 1996); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP-3 (Farrugia, 1997) and PLATON (Spek, 1990)'; software used to prepare material for publication: SHELXL97 and PARST (Nardelli, 1995).

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